

B

**STANDARD STATE
THERMODYNAMIC
PROPERTIES OF SELECTED
MINERALS AND OTHER
COMPOUNDS**

Part 1. Inorganic Substances

Data from Wagman et al., The NBS tables of chemical thermodynamic properties. Jour. Physical and Chemical Reference Data, v. 11, Supplement no. 2, 1982; with a few additions from other sources—Al species from Drever (1988); silica species and all volume data from SUPCRT92 (Johnson et al., 1992).

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$ kJ mol ⁻¹	$\Delta_f G^\circ$ kJ mol ⁻¹	S° J mol ⁻¹ K ⁻¹	C_p° J mol ⁻¹ K ⁻¹	V° cm ³ mol ⁻¹
Aluminum							
Al	s	26.9815	0	0	28.33	24.35	
Al ³⁺	aq	26.9815	-531.	-485.	-321.7	—	-45.3
Al(OH) ²⁺	aq		-767.0	-693.7	—	—	
Al(OH) ₂ ⁺	aq		-1010.7	-901.4	—	—	
Al(OH) ₃ ⁰ (aq)	aq		-1250.4	-1100.7	—	—	
Al(OH) ₄ ⁻	aq	95.0111	-1490.0	-1307.0	102.9	—	45.60
Al ₂ O ₃	α , corundum	101.9612	-1675.7	-1582.3	50.92	79.04	25.575
Al ₂ O ₃ · H ₂ O	boehmite	119.9766	-1980.7	-1831.7	96.86	131.25	39.07
Al ₂ O ₃ · H ₂ O	diaspore	119.9766	-1998.91	-1841.78	70.67	106.19	35.52
Al ₂ O ₃ · 3H ₂ O	gibbsite	156.0074	-2586.67	-2310.21	136.90	183.47	63.912
Al ₂ O ₃ · 3H ₂ O	bayerite	156.0074	-2576.5	—	—	—	
Al(OH) ₃	amorphous	78.0037	-1276.	—	—	—	
Al ₂ SiO ₅	andalusite	162.0460	-2590.27	-2442.66	93.22	122.72	51.53
Al ₂ SiO ₅	kyanite	162.0460	-2594.29	-2443.88	83.81	121.71	44.09
Al ₂ SiO ₅	sillimanite	162.0460	-2587.76	-2440.99	96.11	124.52	49.90
Al ₂ Si ₂ O ₇ · 2H ₂ O	kaolinite	258.1616	-4119.6	-3799.7	205.0	246.14	99.52
Al ₂ Si ₂ O ₇ · 2H ₂ O	halloysite	258.1616	-4101.2	-3780.5	203.3	246.27	99.30
Al ₂ Si ₂ O ₇ · 2H ₂ O	dickite	258.1616	-4118.3	-3795.9	197.1	239.49	99.30
Al ₆ Si ₂ O ₁₃	mullite	426.0532	-6816.2	-6432.7	255.	326.10	-
Al ₂ Si ₄ O ₁₀ (OH) ₂	pyrophyllite	360.3158	-5642.04	-5268.14	239.41	294.34	126.6
Barium							
Ba	s	137.3400	0	0	62.8	28.07	
Ba ²⁺	aq	137.3400	-537.64	-560.77	9.6	—	-12.9
BaO	s	153.3394	-553.5	-525.1	70.42	47.78	
BaO ₂	s	169.3388	-634.3	—	—	66.9	
BaF ₂	s	175.3368	-1207.1	-1156.8	96.36	71.21	
BaS	s	169.4040	-460.	-456.	78.2	49.37	
BaSO ₄	barite	233.4016	-1473.2	-1362.2	132.2	101.75	52.10
BaCO ₃	witherite	197.3494	-1216.3	-1137.6	112.1	85.35	45.81
BaSiO ₃	s	213.4242	-1623.60	-1540.21	109.6	90.00	
Calcium							
Ca	s	40.0800	0	0	41.42	25.31	
Ca ²⁺	aq	40.0800	-542.83	-553.58	-53.1	—	-18.4
CaO	s	56.0794	-635.09	-604.03	39.75	42.80	
Ca(OH) ₂	portlandite	74.0948	-986.09	-898.49	83.39	87.49	
CaF ₂	fluorite	78.0768	-1219.6	-1167.3	68.87	67.03	24.542
CaS	s	72.1440	-482.4	-477.4	56.5	47.40	
CaSO ₄	anhydrite	136.1416	-1434.11	-1321.79	106.7	99.66	45.94

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$ kJ mol ⁻¹	$\Delta_f G^\circ$ kJ mol ⁻¹	S° J mol ⁻¹ K ⁻¹	C_p° J mol ⁻¹ K ⁻¹	V° cm ³ mol ⁻¹
CaSO ₄ · 2H ₂ O	gypsum	172.1724	-2022.63	-1797.28	194.1	186.02	
Ca ₃ (PO ₄) ₂	β , whitlockite	310.1828	-4120.8	-3884.7	236.0	227.82	
Ca ₃ (PO ₄) ₂	α	310.1828	-4109.9	-3875.5	240.91	231.58	
CaCO ₃	calcite	100.0894	-1206.92	-1128.79	92.9	81.88	36.934
CaCO ₃	aragonite	100.0894	-1207.13	-1127.75	88.7	81.25	34.150
CaSiO ₃	wollastonite	116.1642	-1634.94	-1549.66	81.92	85.27	39.93
CaSiO ₃	pseudowollastonite	116.1642	-1628.4	-1544.7	87.36	86.48	
CaAl ₂ SiO ₆	Ca-Al pyroxene	218.1254	-3298.2	-3122.0	141.4	165.7	
CaAl ₂ Si ₂ O ₈	anorthite	278.2102	-4227.9	-4002.3	199.28	211.42	100.79
CaTiO ₃	perovskite	135.9782	-1660.6	-1575.2	93.64	97.65	
CaTiSiO ₅	sphene	196.0630	-2603.3	-2461.8	129.20	138.95	
CaMg(CO ₃) ₂	dolomite	184.4108	-2326.3	-2163.4	155.18	157.53	64.365
CaMgSi ₂ O ₆	diopside	216.5604	-3206.2	-3032.0	142.93	166.52	66.090
Carbon							
C	graphite	12.0112	0	0	5.740	8.527	5.298
C	diamond	12.0112	1.895	2.900	2.377	6.113	3.417
CO ₃ ²⁻	aq	60.0094	-677.149	-527.81	-56.9	—	-6.1
HCO ₃ ⁻	aq	61.0174	-691.99	-586.77	91.2	—	24.2
CO	g	28.0106	-110.525	-137.168	197.674	29.142	24465.6
CO ₂	g	44.0100	-393.509	-394.359	213.74	37.11	24465.6
CO ₂	aq	44.0100	-413.80	-385.98	117.6	—	32.8
H ₂ CO ₃	aq	62.0254	-679.339	-623.109	283.65	—	
CH ₄	g	16.0432	-74.81	-50.72	186.264	35.309	24465.6
C ₂ H ₆	g	30.0704	-84.68	-32.82	229.60	52.63	24465.6
CN	g	26.0179	437.6	407.5	202.6	29.16	
CN ⁻	aq	26.0179	150.6	172.4	94.1	—	
HCN	g	27.0259	135.1	124.7	201.78	35.86	
HCN	aq	27.0259	107.1	119.7	124.7	—	
Chlorine							
Cl ₂	g	70.9060	0	0	233.066	33.907	24465.6
Cl ⁻	aq	35.4530	-167.159	-131.228	56.5	-136.4	17.3
HCl	aq	36.4610	-167.159	-131.228	56.5	-136.4	17.3
HCl	g	36.4610	-92.307	-95.299	186.908	29.12	24465.6
Copper							
Cu	s	63.5400	0	0	33.15	24.435	
Cu ⁺	aq	63.5400	71.67	49.98	40.6	—	
Cu ²⁺	aq	63.5400	64.77	65.49	-99.6	—	
CuO	tenorite	79.5394	-157.3	-129.7	42.63	42.30	
Cu ₂ O	cuprite	143.0794	-168.6	-146.0	93.14	63.64	
CuS	covellite	96.6040	-53.1	-53.6	66.5	47.82	
Cu ₂ S	chalcocite	159.1440	-79.5	-86.2	120.9	76.32	

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$	$\Delta_f G^\circ$	S°	C_p°	V°
				kJ mol ⁻¹	J mol ⁻¹ K ⁻¹		cm ³ mol ⁻¹
Fluorine							
F ₂	g	37.9968	0	0	202.78	31.30	
HF	g	20.0064	-271.1	-273.2	173.779	29.133	
HF	aq	20.0064	-320.08	-296.82	88.7	—	
F ⁻	18.9984	-332.63	-278.79	-13.8	-106.7	—	
Hydrogen							
H ₂	g	2.0160	0	0	130.684	28.824	24465.6
H ⁺	aq	1.0080	0	0	0	0	0
OH ⁻	aq	17.0074	-229.994	-157.244	-10.75	-148.5	
H ₂ O	l	18.0154	-285.830	-237.129	69.91	75.291	18.068
H ₂ O	g	18.0154	-241.818	-228.572	188.825	33.577	24465.6
Iodine							
I ₂	s	253.8088	0	0	116.135	54.438	
I ⁻	aq	126.9044	-55.19	-51.57	111.3	-142.3	
HI	aq	127.9124	-55.19	-51.57	111.3	—	
IO ₃ ⁻	aq	174.9026	-221.3	-128.0	118.4	—	
IO ₄ ⁻	aq	190.9020	-155.5	-58.5	222.0	—	
Iron							
Fe	s	55.8470	0	0	27.28	25.10	
Fe ²⁺	aq	55.8470	-89.1	-78.90	-137.7	—	
Fe ³⁺	aq	55.8470	-48.5	-4.7	-315.9	—	
Fe _{0.947} O	wüstite	68.8865	-266.27	-245.12	57.49	48.12	
Fe ₂ O ₃	hematite	159.6922	-824.2	-742.2	87.40	103.85	
Fe ₃ O ₄	magnetite	231.5386	-1118.4	-1015.4	146.4	143.43	
FeO(OH)	goethite	88.8538	-559.0	(-487.02)	(60.25)	—	
Fe(OH) ₂	s	89.8618	-569.0	-486.5	88.	—	
Fe(OH) ₃	s	106.8692	-823.0	-696.5	106.7	—	
FeS	troilite	87.9110	-100.0	-100.4	60.29	50.54	
FeS ₂	pyrite	119.9750	-178.2	-166.9	52.93	62.17	
FeCO ₃	siderite	115.8564	-740.57	-666.67	92.9	82.13	
Fe ₂ SiO ₄	fayalite	203.7776	-1479.9	-1379.0	145.2	132.88	
Lead							
Pb	s	207.1900	0	0	64.81	26.44	
Pb ²⁺	aq	207.1900	-1.7	-24.43	10.5	—	
PbO	yellow	223.1894	-217.32	-187.89	68.70	45.77	
PbO	red	223.1894	-218.99	-188.93	66.5	45.81	

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$	$\Delta_f G^\circ$	S°	C_p°	V°
				kJ mol ⁻¹	J mol ⁻¹ K ⁻¹	J mol ⁻¹ K ⁻¹	cm ³ mol ⁻¹
PbF ₂	s	245.1868	-664.0	-617.1	110.5	—	
PbCl ₂	s	278.0960	-359.41	-314.10	136.0	—	
PbS	galena	239.2540	-100.42	-98.7	91.2	49.50	
PbSO ₄	anglesite	303.2516	-919.94	-813.14	148.57	103.207	
PbCO ₃	cerussite	267.1994	-699.1	-625.5	131.0	87.40	
PbSiO ₃	s	283.2742	-1145.70	-1062.10	109.6	90.04	
Magnesium							
Mg	s	24.3120	0	0	32.68	24.89	
Mg ²⁺	aq	24.3120	-466.85	-454.8	-138.1	—	
MgO	periclase	40.3114	-601.70	-569.43	26.94	37.15	
Mg(OH) ₂	brucite	58.3268	-924.54	-833.51	63.18	77.03	
MgF ₂	sellaite	62.3088	-1123.4	-1070.2	57.24	61.59	
MgS	s	56.3760	-346.0	-341.8	50.33	45.56	
MgCO ₃	magnesite	84.3214	-1095.8	-1012.1	65.7	75.52	28.018
MgCO ₃ · 3H ₂ O	nesquehonite	138.3676	—	-1726.1	—	—	
MgSiO ₃	enstatite	100.3962	-1549.00	-1462.09	67.74	81.38	
Mg ₂ SiO ₄	forsterite	140.7076	-2174.0	-2055.1	95.14	118.49	
Manganese							
Mn	s	54.9380	0	0	32.01	26.32	
Mn ²⁺	aq	54.9380	-220.75	-228.1	-73.6	50.	
MnO ₄ ⁻	aq	118.9356	-541.4	-447.2	191.2	-82.0	
MnO	manganosite	70.9374	-385.22	-362.90	59.71	45.44	
Mn ₃ O ₄	hausmannite	228.8116	-1387.8	-1283.2	155.6	139.66	
Mn ₂ O ₃	s	157.8742	-959.0	-881.1	110.5	107.65	
MnO ₂	pyrolusite	86.9368	-520.03	-465.14	53.05	54.14	
Mn(OH) ₂	amorphous	88.9528	-695.4	-615.0	99.2	—	
MnS	alabandite	87.0020	-214.2	-218.4	78.2	49.96	
MnCO ₃	rhodochrosite	114.9474	-894.1	-816.7	85.8	81.50	
MnSiO ₃	rhodonite	131.0222	-1320.9	-1240.5	89.1	86.44	
Mn ₂ SiO ₄	tephroite	201.9596	-1730.5	-1632.1	163.2	129.87	
Mercury							
Hg	l	200.5900	0	0	76.02	27.983	
Hg	g	200.5900	61.317	31.820	174.96	20.786	
Hg ²⁺	aq	200.5900	171.1	164.4	-32.2	—	
Hg ₂ ²⁺	aq	401.1800	172.4	153.52	84.5	—	
HgS ₂ ²⁻	aq	264.7180	—	41.9	—	—	
HgCl ₄ ²⁻	aq	342.4020	-554.0	-446.8	293.	—	
Hg ₂ Cl ₂	s	472.0860	-265.22	-210.745	192.5	—	
HgO	s, red	216.5894	-90.83	-58.539	70.29	44.06	
HgO	s, yellow	216.5894	-90.46	-58.409	71.1	—	
HgS	cinnabar	232.6540	-58.2	-50.6	82.4	48.41	
HgS	metacinnabar	232.6540	-53.6	-47.7	88.3	—	

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$ kJ mol ⁻¹	$\Delta_f G^\circ$ kJ mol ⁻¹	S° J mol ⁻¹ K ⁻¹	C_p° J mol ⁻¹ K ⁻¹	V° cm ³ mol ⁻¹
Molybdenum							
Mo	s	95.9400	0	0	28.66	24.06	
MoO ₃	s	127.9388	-745.09	-667.97	77.74	74.98	
MoS ₂	molybdenite	160.0680	-235.1	-225.9	62.59	63.55	
Nickel							
Ni	s	58.7100	0	0	29.87	26.07	
Ni ²⁺	aq	58.7100	-54.0	-45.6	-128.9	—	
NiO	bunsenite	74.7094	-239.7	-211.7	37.99	44.31	
NiS	s	90.7740	-82.0	-79.5	52.97	47.11	
Nitrogen							
N ₂	g	28.0134	0	0	191.61	29.125	
NO	g	30.0061	90.25	86.55	210.761	29.844	
NO ₂	g	46.0055	33.18	51.31	240.06	37.20	
N ₂ O	g	44.0128	82.05	104.2	219.85	38.45	
N ₂ O ₄	l	92.0110	-19.50	97.54	209.2	142.7	
N ₂ O ₄	g	92.0110	9.16	97.89	304.29	77.28	
N ₂ O ₅	s	108.0104	-43.1	113.9	178.2	143.1	
N ₂ O ₅	g	108.0104	11.3	115.1	355.7	84.5	
NH ₃	g	17.0307	-46.11	-16.45	192.45	35.06	
NO ₃ ⁻	aq	62.0049	-205.0	-108.74	146.45	-86.6	
NH ₄ ⁺	aq	18.0837	-132.51	-79.31	113.4	79.9	
NH ₄ OH	aq	35.0461	-366.12	-263.63	181.21	—	
Oxygen							
O ₂	g	31.9988	0	0	205.138	29.355	
O ₂	aq	31.9988	-11.7	16.4	110.9	—	
OH ⁻	aq	17.0074	-229.994	-157.244	-10.75	-148.5	
H ₂ O	l	18.0154	-285.830	-237.129	69.91	75.291	18.068
H ₂ O	g	18.0154	-241.818	-228.572	188.825	33.577	24465.6
Potassium							
K	s	39.1020	0	0	64.18	29.58	
K ⁺	aq	39.1020	-252.38	-283.27	102.5	21.8	9.0
KCl	sylvite	74.5550	-436.747	-409.14	82.59	51.30	
KAlSi ₃ O ₈	sanidine	278.3367	-3959.7	-3739.9	232.88	204.51	
KAlSi ₃ O ₈	microcline	278.3367	-3968.1	-3742.9	214.22	202.38	108.741
KAlSiO ₄	kaliophilite	158.1671	-2121.3	-2005.3	133.1	119.79	
KAlSi ₂ O ₆	leucite	218.2519	-3034.2	-2871.4	200.08	164.14	
KAl ₃ Si ₃ O ₁₀ OH ₂	muscovite	398.3133	-5984.4	-5608.4	306.3	—	14.087

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$	$\Delta_f G^\circ$	S°	C_p°	V°
				kJ mol ⁻¹	J mol ⁻¹ K ⁻¹	J mol ⁻¹ K ⁻¹	cm ³ mol ⁻¹
Silicon							
Si	s	28.0860	0	0	18.83	20.00	
SiO ₂	α -quartz	60.0848	-910.94	-856.64	41.84	44.43	22.688
SiO ₂	α -cristobalite	60.0848	-909.48	-855.43	42.68	44.18	
SiO ₂	α -tridymite	60.0848	-909.06	-855.26	43.5	44.60	25.740
SiO ₂	coesite	60.0848	-906.31	-851.62	40.376	43.51	20.641
SiO ₂	amorphous	60.0848	-903.49	-850.70	46.9	44.4	
SiO ₂	aq	60.0848	-877.699	-833.411	75.312	318.40	16.1
H ₄ SiO ₄	aq		-1449.359	-1307.669	215.132	468.98	
HSiO ₃ ⁻	aq		-1125.583	-1013.783	41.84	-137.24	9.5
Silver							
Ag	s	107.8700	0	0	42.55	25.351	
Ag ⁺	aq	107.8700	105.579	77.107	72.68	21.8	
Ag ₂ O	s	231.7394	-31.05	-11.20	121.3	65.86	
AgCl	cerargyrite	143.3230	-127.068	-109.789	96.2	50.79	
Ag ₂ S	acanthite	247.8040	-32.59	-40.67	144.01	76.53	
Ag ₂ S	argentite	247.8040	-29.41	-39.46	150.6	—	
Sodium							
Na	s	22.9898	0	0	51.21	28.24	
Na ⁺	aq	22.9898	-240.12	-261.905	59.0	46.4	-1.2
NaCl	halite	58.4428	-411.153	-384.138	72.13	50.50	27.015
Na ₂ SiO ₃	s	122.0638	-1554.90	-1462.80	113.85	—	
NaAlSiO ₄	nepheline	142.0549	-2092.8	-1978.1	124.3	—	54.16
NaAlSi ₃ O ₈	low albite	262.2245	-3935.1	-3711.5	207.40	205.10	100.07
NaAlSi ₂ O ₆	jadeite	202.1397	-3030.9	-2852.1	133.5	—	60.40
Sulfur							
S	orthorhombic	32.0640	0	0	31.80	22.64	
S ²⁻	aq	32.0640	33.1	85.8	-14.6	—	
HS ⁻	aq	33.0720	-17.6	12.08	62.8	—	
SO ₄ ²⁻	aq	96.0616	-909.27	-744.53	20.1	-293.	
HSO ₄ ⁻	aq			-756.0		—	
S ₂	g	64.1280	128.37	79.30	228.18	32.47	
H ₂ S	g	34.0800	-20.63	-33.56	205.79	34.23	
H ₂ S	aq	34.0800	-39.7	-27.83	121.	—	
SO ₂	g	64.0628	-296.830	-300.194	248.22	39.87	
SO ₃	g	80.0622	-395.72	-371.06	256.76	50.67	

Formula	Form	Mol. wt. g mol ⁻¹	$\Delta_f H^\circ$	$\Delta_f G^\circ$	S°	C_p°	V°
				kJ mol ⁻¹	J mol ⁻¹ K ⁻¹	K ⁻¹	cm ³ mol ⁻¹
Titanium							
Ti	s	47.9000	0	0	30.63	25.02	
TiO	s	63.8994	-519.7	-495.0	50.0	39.96	
TiO ₂	anatase	79.8988	-939.7	-884.5	49.92	55.48	
TiO ₂	brookite	79.8988	-941.8	—	—	—	
TiO ₂	rutile	79.8988	-944.7	-889.5	50.33	55.02	
Uranium							
U	s	238.0290	0	0	50.21	27.665	
UO ₂	uraninite	270.0278	-1084.9	-1031.7	77.03	63.60	
UO ₃	orthorhombic	286.0272	-1223.8	-1145.9	96.11	81.67	
U ³⁺	aq	238.0290	-489.1	-475.4	192.	—	
U ⁴⁺	aq	238.0290	-591.2	-531.0	410.	—	
UO ₂ ²⁺	aq	270.0278	-1019.6	-953.5	-97.5	—	
Zinc							
Zn	s	65.3700	0	0	41.63	25.40	
Zn ²⁺	aq	65.3700	-155.89	-147.06	-112.1	46.	
ZnO	zincite	81.3694	-348.28	-318.30	43.64	40.25	
ZnS	wurtzite	97.4340	-192.63	—	—	—	
ZnS	sphalerite	97.4340	-205.98	-201.29	57.7	46.0	
ZnCO ₃	smithsonite	125.3794	-812.78	-731.52	82.4	79.71	
Zn ₂ SiO ₄	willemite	222.8236	-1636.74	-1523.16	131.4	123.34	